Oligomer Structure Effect on HDT in 3D Printing



Heather Francis Ahmet Nebioglu Bomar, Torrington, CT USA



Abstract

The present work addresses a strong market demand for oligomers for 3D printing formulations. Structure–property relationships in a new family of oligomers will be discussed. Effect of oligomers on oxygen inhibition, coefficient of thermal expansion, heat distortion temperature, shrinkage, and color stability will be revealed. Some of the formulations based on the various new oligomers are very fast curing and prone to substantially less oxygen inhibition. They demonstrate good adhesion to different substrates and possess very low water absorption characteristics.

Introduction

3D printing uses additive manufacturing processes that use gradual creation or addition of materials to form an object. 3D printing has become a widely used method for manufacturing prototypes in many industries, including automotive, aerospace, consumer goods, and medical device, since it produces prototypes with complex shapes relatively quickly and at a significantly lower cost than traditional methods.

There are a number of different 3D printing methods but the most common 3D printing methods that use light to form the desired 3D shapes are stereolithography (SLA), UV inkjet, and digital light processing (DLP).

- SLA SLA uses a UV laser to crosslink a liquid resin in a specific area with a specific depth. Once the first layer of the desired object is formed, it is lowered and the liquid resin is re-applied on top of it. Successive curing of layers will form the desired object with high precision.
- UV Inkjet The 3D UV inkjet printing method works similar to a 2D UV inkjet printer. Instead of printing on paper, successive layers are printed and cured on top of the previous layer.
- DLP DLP uses a digital light projector to project a 2D image and cure the resin. Successive curing of the 2D images form the 3D shape desired. Compared to the other methods, the light irradiance used in DLP is significantly lower.

Replacing traditional mass manufacturing processes with 3D printing is very appealing from a commercial perspective, especially in applications where only relatively small number of parts are required. Unfortunately, there are several challenges preventing the realization of this

potential; speed of manufacturing and meeting the material performance properties are the most important ones. From a materials perspective, it has been a challenge to match the performance of common plastics like PC and ABS. It is often difficult to match deformation resistance of plastics at elevated temperatures with thermoset materials. In this study we investigated oligomers used in the 3D printing industry and how to improve their heat distortion temperatures without having to significantly take away from other performance attributes.

Experimental

All oligomers were analyzed with gel permeation chromatography (Polymer Labs PL-GPC 50 equipped with RI detector and mixed D-columns and FTIR (Perkin Elmer Spectrum 100). Viscosities were measured using a Brookfield CAP 2000+ viscometer. The oligomers were tested for potential use in 3D printing applications using a simple model formula (Table 1). Formulations were mixed using a FlackTek DAC 150.1 FVZ SpeedMixer at 3,000 rpm for 2.5 minutes. For ease of testing and comparison, formulations were cast in single layer films where applicable. Such films could be viewed as corresponding to the individually cured layers of a fully 3D printed object.

Table 1. Model Formula

Component	Wt
(Meth)acrylate monomer	51.1%
Benzoylformic acid methyl ester (MBF)	1.9%
Benzophenone	1.7%
(2,4,6-Trimethylbenzoyl)diphenylphosphine oxide (TPO)	0.3%
Oligomer	45.0%

Each formulation was cured with a Dymax UVCS Light—Curing Conveyor outfitted with 5000–EC Flood Lamps at an irradiance of 300 mW/cm² and 4,000 mJ/cm² energy. The curing irradiance and energy were recorded using an ACCU–CAL $^{\infty}$ 150 Radiometer.

The durometer hardness of each formulation was measured using ASTM D2240. Cure profiles of all formulations were analyzed by a Differential Scanning Calorimeter (DSC) equipped with a Dymax BlueWave® 200 Spot Lamp. After 30 seconds of isothermal stabilization at 25°C, samples

were exposed to 50 mW/cm² UV irradiance for 15 seconds. Tensile properties were measured with an Instron according to ASTM D882. Themal Analysis DMA Q800 was used to measure the heat deflection temperature (ASTM D648) and the glass transition temperature (ASTM E1640).

Results and Discussion

In general, polyurethane acrylates are used in 3D printing formulations primarily due to the toughness they provide. This toughness is critical in obtaining impact and scratch-resistant objects. Three different types of polyurethane acrylates were synthesized in order to investigate how the structure of polyurethane acrylates affects the critical properties needed for 3D printing resins. Representative structures of these polyurethane acrylate (PUA) oligomers are given in Figure 1. The first type is the most common type of PUA. The backbone contains soft and hard segments and acrylate groups are connected to the hard segment. In the second type, the acrylate group is directly connected to the soft segment. The third type of PUA has 3 acrylate functionality.

the soft and hard segments of each oligomer respectively compared to other oligomers. According to Table 1, oligomers 1B, 1C, and 1D have the same amount of HSR. With an increase in SSR, the viscosity of the formulations decreased.

Table 2 lists the physical properties of the cured formulations. Heat distortion temperature (HDT) is important for operating temperature of the printed objects, especially for non-prototype applications. Oligomers 1C and 1D gave

Table 1 shows the Type 1 oligomers' structural description

and viscosity of the model formula with the corresponding

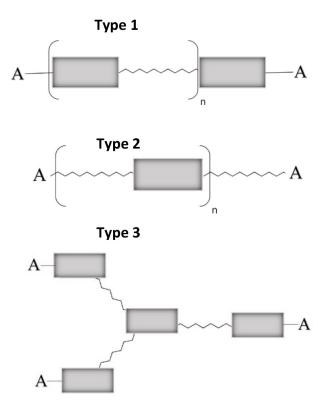
acrylate groups within each oligomer. Soft segment ratio

(SSR) and hard segment ratio (HSR) are relative length of

oligomer. Acrylate ratio is defined as the relative ratio of the

for operating temperature of the printed objects, especially for non-prototype applications. Oligomers 1C and 1D gave the highest HDT. Both of them had high HSR and the highest acrylate amount (AR). Oligomer 1D had higher SSR, which resulted in a slightly higher elongation at break and a lower tensile modulus compared to 1C. It was interesting that the HDT of 1D was same as 1C even though it had higher SSR. From 1B to 1C, AR and SSR were increased by about the same factor, as a result the HDT increased (despite the increase in SSR). 1A provided very good elongation due to its high SSR. It

Figure 1. Different Types Polyurethane Acrylates



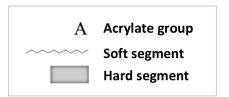


Table 1. Oligomers with Type 1 Structure

Oligomer	Acrylate Functionality	Acrylate Ratio (AR)	Soft Segment Ratio (SSR)	Hard Segment Ratio (HSR)	Viscosity (cP)
1A	2	1.0	67.7	1	850
1B	2	8.9	1.7	7	1,000
1C	2	12.1	2.6	7	680
1D	2	12.1	9.0	7	430

Table 2. Cured Properties of the Model Formula with Type 1 Structured Oligomers

	Tensile Strength (psi)	Elongation (%)	Modulus (MPa)	HDT (°C)	Tg (°C)	Durometer Hardness (D)	Linear Shrinkage (%)	Water Absorption (%)
1A	2,000	550	140	36	31	29	0.8	0.48
1B	4,200	2	900	110	131	89	0.6	0.16
1C	4,400	2.7	690	124	143	86	0.87	0.2
1D	4,400	3.5	620	124	145	86	0.95	0.15

Table 3. Oligomers with Type 2 Structure

	Acrylate Functionality	Acrylate Ratio	Soft Segment Ratio	Hard Segment Ratio	Viscosity (cP)
2A	2	8.7	1.1	11	174
2B	2	9.4	1.0	12	191
2C	2	6.8	1.6	10	88

Table 4. Cured Properties of the Model Formula with Type 2 Structured Oligomers

	Tensile Strength (psi)	Elongation (%)	Modulus (MPa)	HDT (oC)	Tg (oC)	Durometer Hardness (D)	Linear shrink- age (%)	Water Absorp- tion (%)
2A	7,200	6	690	74	85	82	0.4	0.18
2B	6,200	5	620	80	96	83	0.46	0.17
2C	5,100	3.2	690	72	88	77	0.55	0.19

Table 5. Oligomers with Type 3 Structure

	Acrylate Functionality	Acrylate Ratio (AR)	Soft Segment Ratio (SSR)	Hard Segment Ratio (HSR)	Viscosity (cP)
3A	3	7.6	4.1	7	949
3B	3	7.1	0.9	24	597
3C	3	5.9	1.3	21	541

can be used as an additive oligomer in objects that require relatively high elongation rates.

Table 3 shows the Type 2 oligomers' structural description and viscosity of the model formula with the corresponding oligomer. Type 2 oligomers provided significantly lower viscosities compared to Type 1 oligomers. The viscosity of the formulation with oligomer 2C was exceptionally low.

Type 2 oligomers provided lower HDT compared relatively similar SSR, HSR, and AR containing Type 1 oligomers. This might be related to hard segments' proximity to acrylate groups and hence hard segments' proximity to crosslinks in the cured material. When the hard segments are closer to crosslinks it might be harder to distort the crosslinks and therefore result in higher HDT. Type 2 oligomers also provided higher tensile strength and lower linear shrinkage values compared to Type 1 oligomers. Lower shrinkage might be due to relatively less stress formation during curing due to hard segments' proximity to the crosslinking site. Less stress build up in the material also affects tensile properties. An increase of HSR and AR and a decrease in SSR seem to increase HDT. Oligomer 2A is appealing due to very low shrinkage and good balance of tensile strength and elongation.

Oligomers with Type 3 structure have average of three acrylate functionalities per polymer chain. It does not seem like increasing functionality of an oligomer simply improves HDT. When compared to 1B, 3A has the same HSR and also has relatively higher SSR and lower AR, which as expected, result in lower HDT than 1B. Higher functionality of 3A did not improve HDT. Oligomers 3B and 3C have significantly higher HSR and lower SSR than 3A, which resulted into higher HDT. It is interesting to note that 3A has practically the same Tg as 3B and 3C, but it has significantly lower HDT. This is most likely due to higher amount of trapped soft segments that are relatively easier to deform at elevated temperatures.

The cure speed of a formulation is an important factor in many 3D printing applications. Relative cure speeds of the formulations were measured by photo–DSC, based on time to reach gel point. Formulations with Type 2 oligomers cured relatively slowly. Type 1 oligomers cured relatively fast except very high SSR ratio oligomer (1A). Among Type 3 oligomers, 3B cured fastest possibly due to a lower SSR ratio.

Table 6. Cured Properties of the Model Formula with Type 3 Structured Oligomer

	Tensile Strength (psi)	Elongation (%)	Modulus (MPa)	HDT (°C)	Τ _g (°C)	Durometer Hardness (D)	Linear Shrinkage (%)	Water Absorption (%)
3A	5,000	3	760	70	118	87	0.7	0.22
3B	6,000	3.2	760	99	119	87	0.64	0.18
3C	4,000	2.7	550	93	114	84	0.67	0.2

Table 7. Relative Cure Speeds (Photo-DSC)

	1A	1B	10	1D	2A	2B	2C	3A	3B	3C
Relative Cure Speed	1.5	2.8	3.0	2.2	1.5	1.4	1.4	1.2	2.0	1.5

Conclusion

In this paper, we have described several, new polyurethane acrylate oligomers of differing structure which may find uses in a wide range of 3D printing applications. These oligomers' physical properties have been compared in a model formula and various structure—property relations have been explored. Several oligomers exhibited high heat distortion temperatures and fast cure speeds. Such oligomers could be useful in applications where the material properties need to more closely resemble common plastics like PC or ABS. Still other oligomers may have value in different 3D printing applications requiring a material with a balance of hardness, toughness, and flexibility. Understanding the relationship between the structure of these oligomers and their ultimate properties can lead the way to tailor—made products for each 3D printing application.

References

- "The Ultimate Guide to Stereolithography (SLA) 3D Printing." Formlabs.com. March 2017, pg. 6. Available for download at https://formlabs.com/blog/ultimate_guide_to_stereolithography_sla_3d_printing/.
- 2. See e.g. Francis, Heather and A. Nebioglu. "New Oligomers for 3D Printing Inks." UV+EB Technology. Issue 1, 2016. Available at http://www.uvebtech.com/sto-ries/022616/3d-printing.shtml#.WrQqMGffPIU.

Global Headquarters: 51 Greenwoods Road | Torrington, CT 06790 | USA | +1860-626-7006

www.bomar-chem.com